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## **Interpreting Graphics Chemistry Answers 17**

494 CHAPTER 17 Chemical Reactions Chemical Equations To describe a chemical reaction, you must know which substances react and which substances are formed in the reaction. The substances that react are called the reactants (see AK tunts). Reactants are the substances that exist before the reaction begins.

## **649-S1-MSS05 8/13/04 4:54 PM Page 490 Chemical Reactions**

Chapter 17 "Thermochemistry" Tools. Copy this to my account; E-mail to a friend; Find other activities; Start over; Help; Use these activities to learn the vocabulary and major concepts presented in this chapter. A B; calorie: the quantity of heat that raises the temperature of 1 g of pure water by 1°C:

## **Quia - Chapter 17 "Thermochemistry"**

Fig. 17.2, page 506 - on the left, the system (the people) gain heat from its surroundings (the fire) this is . endothermic (q is positive) On the right, the system (the body) cools as perspiration evaporates, and heat flows to the surroundings. this is . exothermic (q is negative)

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organisms that eat squid. 4. How many producers are in the food  
web? Name them. Use the figures below, which show trophic  
levels in an ecosystem, to complete items 5 and ...

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Calorimeter . N/A . N/A : Hot Water . Calculations .  $q_{\text{water}} + q_{\text{calorimeter}} + q_{\text{reaction}} = 0$  ...

## Heat Transfer: Counting Calories Activity - Your ...

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