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a heterogeneous

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mixture of intermediate-sized particles (between atomic-size of solution particles and the size of suspension particles) Brownian motion the erratic, random, movements of colloid particles that results from collisions of particles of the dispersion medium with the dispersed particles

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**(Mixtures and
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Chemistry II: Chapter
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Solutions. a type of heterogeneous mixture whose particles settle out over time and can be separated from the mixture by filtration. a heterogeneous mixture of intermediate-sized particles (between atomic-size of solution particles and the size of suspension

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Mixtures

Chemistry II: Chapter 14 Mixtures and Solutions ...

The Mixtures and Solutions chapter of this Glencoe Chemistry - Matter and Change companion course helps students learn the essential chemistry lessons of mixtures, solvents and solutes.

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Heterogeneous and

Homogeneous

Mixtures. 1. false. 2.

true. 3. false. 4. true. 5.

true. 6. false. 7. true. 8.

true. 9. false. 10. true.

Characteristics of

Particles Colloid

Solution Suspension

11. Less than 1 nm in

diameter (12. Between

1 nm and 1000 nm in

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diameter (13. More
than 1000 nm in
diameter (14. Settle
out if undisturbed (15.

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A mixture is matter that is made up of

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atoms of two or more
elements chemically
bonded together.

Physical Science - Chapter 13-Mixtures, Solubility, and ...

Tyndall effect soluble..

3 Evaluate the

Answer.Start studying
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study guide: mixtures
and solutions. If one
liquid is soluble in
another liquid, such as
acetic acid in water,

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the two liquids are

Solvation is the process of surrounding solute particles with solvent particles to

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important use of
colligative properties is
to determine the molar
mass of unknown ...

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