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CHAPTER 11

SOLUTIONS MANUAL c.

How many moles of H_2S are produced? 1.50

mol S_8 4 mol H_2S 1

mol S_8 6.00 mol H_2S

12. Challenge Sulfuric

acid (H_2SO_4) is

formed when sulfur

dioxide (SO_2) reacts

with oxygen and water.

a. Write the balanced
chemical equation for

the reaction. $2\text{SO}_2(\text{g})$

$\text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{SO}_4(\text{l})$

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SOLUTIONS MANUAL

10. Explain how a mole
is similar to a dozen.

The mole is a unit for
counting 6.02 10²³

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representative
particles. The dozen is
used to count 12 items.

11. Apply How does a
chemist count the
number of particles in
a given number of
moles of a substance?

The MoleThe Mole

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